# Synthetic Studies of Withanolides. Part I. ${ }^{1}$ Synthesis of 5,6 $\beta$-Epoxy-4 $\beta$ -hydroxy-5 $\beta$-cholest-2-en-1-one and Related Compounds 

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The title compounds, which have functionalities in rings $A$ and $B$ similar to those of withaferin $A(1)$, have been synthesized through the key intermediates cholesta-2,5-dien-1-one (10) and -2,4-dien-1-one (26). Electrophilic reactions of the steroidal 2,4 -dien-1-one system were observed to occur stereoselectively on the $\beta$-side of the 4.5double bond.

We have briefly reported ${ }^{1}$ a stereoselective synthesis of $5,6 \beta$-epoxy- $4 \beta$-hydroxy- $5 \beta$-cholest- 2 -en-1-one
(25),

(1)
which has the same functionalities in rings $A$ and $B$ as withaferin $A$ (l), a cytotoxic principle found in some
prompted us to synthesize analogous compounds for elucidating the structure-activity relationship. The present paper gives details of the synthesis of (25) and its analogues (16), (18), (20), (22), (39), and (40), all of which contain an $\alpha$-hydroxy-epoxide system connected to the cholest-2-en-1-one unit. Another congener (30) has been prepared by the method of Weissenberg et al. ${ }^{5}$
The key intermediates were cholesta-2,4-dien-1-one $(26)^{6}$ and its isomer, cholesta-2,5-dien-1-one (10). The latter was prepared by two alternative routes. 6 6 -Acetoxy- $1 \alpha, 2 \alpha$-epoxy- $5 \alpha$-cholestan- 3 -one (2) ${ }^{7}$ was reduced with hydrazine to give the 2 -en- $1 \alpha$-ol (3), which was oxidized with Jones reagent, affording the ketone (4) in $45 \%$ yield from (2). Alkaline hydrolysis of (4) in methanol induced a Michael-type addition yielding the methanol adduct (6), in addition to the expected alcohol


Scheme 1 Reagents: i, $\mathrm{NH}_{2} \cdot \mathrm{NH}_{2}$; ii, Jones reagent; iii, KOH ; iv, $\mathrm{POCl}_{3} ; \mathrm{v}, \mathrm{Ac}_{2} \mathrm{O}-\mathrm{AcOH}$; vi, NaOH
species of Solanaceae. ${ }^{2,3}$ An inhibitory effect of (25) against the growth of Sarcoma 180 ascites tumour ${ }^{4}$

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${ }^{2}$ S. M. Kupchan, R. W. Doskotch, P. Bollinger, A. T. McPhail, G. A. Sim, and J. A. S. Renauld, J. Amer. Chem. Soc., 1965, 87, 5805.
(5). Changing the solvent to dioxan, however, gave the alcohol (5) as the sole product. Dehydration of (5) with ${ }^{3}$ D. Lavie, E. Glotter, and Y. Shovo, J. Chem. Soc., 1965, 7517.
${ }_{4}$ Details of biological activities will be described elsewhere.
${ }^{5}$ M. Weissenberg, E. Glotter, and D. Lavie, Tetrahedron Letters, 1974, 3063.
${ }^{6}$ H. Izawa, M. Morisaki, and K. Tsuda, Chem. and Pharm. Bull. (Japan), 1966, 14, 873.
${ }^{7}$ B. Pelc and E. Kodicek, J. Chem. Soc. (C), 1970, 1624.
phosphoryl chloride in pyridine led to the dienone (10) in $90 \%$ yield.

The second synthesis of ( 10 ) started from $1 \alpha$-hydroxycholesterol (7), the preparation of which has been studied by many groups because of its usefulness as an immediate synthetic precursor of an active form of vitamin $\mathrm{D}_{3}, 1 \alpha-$ hydroxycholecalciferol. ${ }^{8}$ The less hindered $3 \beta$-hydroxygroup in (7) was selectively acetylated with acetic anhydride in acetic acid, giving the monoacetate (8) in $60 \%$ yield ( $39 \%$ of starting material was recovered). Jones
zoic acid afforded a mixture of two epoxides (11) and (12) in the ratio $2: 1$ and these were separated by silica gel column chromatography. Their n.m.r. data (Table) were not adequate for assignment of the respective structures. Treatment of both epoxides (11) and (12) with $3 \%$ perchloric acid in tetrahydrofuran gave the same single glycol (13), suggesting a trans-diaxial ring opening. The configuration of the C-6 hydroxy-group in (13) was established as $\beta$ by its transformation into the $6 \beta$-acetoxydienone (14) by acetylation followed by dehydration.


Scheme 2 Reagents: i, $m$-CPBA; ii, $\mathrm{HClO}_{4}$; iii, $\mathrm{Ac}_{2} \mathrm{O}$; iv, $\mathrm{SOCl}_{2}$; v, MsCl, vi, $\mathrm{NaHCO}_{3}$; vii, NaOH ; viii, $\mathrm{NBA}-\mathrm{AcOLi}$; ix, $\mathrm{OsO} \mathrm{C}_{4}$
oxidation of (8) afforded the ketone (9), and the latter was treated with $1.5 \%$ sodium hydroxide in aqueous dioxan, resulting in elimination of acetic acid to give the dienone ( 10 ) [ $90 \%$ yield from (8)].

Epoxidation of the dienone (10) with $m$-chloroperben-

[^1]The structure of (14) was confirmed by an alternative preparation from the enone (4), involving allylic bromination with $N$-bromosuccinimide and subsequent dehydrobromination with calcium carbonate. On the other hand, the methanesulphonate of the glycol (13a) could be reconverted into the more polar major epoxide (11) by treatment with sodium hydrogen carbonate in pyridine. From those chemical transformations, the major product of epoxidation of ( 10 ) was determined to be the $5 \alpha, 6 \alpha-$ epoxide (11); the $5 \beta, 6 \beta$-epoxide structure (12) was assigned to the less polar minor product.

The $5 \beta, 6 \beta$-epoxide is reported to be the predominant
product of epoxidation by peroxy-acid of withanolide $\mathbf{L}$, which contains a $2,5,14$-trien-1-one system. ${ }^{9}$

When the epoxides (12) and (11) were refluxed with dioxan containing $15 \%$ sodium hydroxide, ${ }^{10}$ the dienone alcohols (15) and (19), respectively, were obtained in nearly quantitative yield. Compound (15) was identified with the hydrolysis product of the acetoxy-dienone (14), confirming its $6 \beta$-hydroxy-configuration. This suggests, in turn, retention of configuration at $\mathrm{C}-6$ of each epoxide during base-catalysed ring openings. ${ }^{11}$ The configurations of (15) and (19) were supported by n.m.r. analysis. Thus, the 10 -methyl signal of (15) appeared at lower field ( $\Delta \delta c a .0 .19$ p.p.m.) than that of (19), indicating the presence of a hydroxy-group in a 1,3 -diaxial relation. Furthermore, the half-height widths of the $\mathrm{C}-6$ proton signals of (15) ( 5 Hz ) and (19) ( 14 Hz ) were compatible with their quasi-equatorial and axial nature, respectively.

Oxidation of compound (15) or (19) with $m$-chloroperbenzoic acid occurred stereoselectively to afford a single epoxide, (16) or (20) respectively. The homogeneity of the products was shown by t.l.c. and n.m.r. spectra. In order to ascertain their stereochemistry, we attempted to prepare the stereoisomeric epoxides. For this purpose, the dienone (15) was treated with $N$-bromoacetamidelithium acetate in acetic acid. ${ }^{\mathbf{1 2}}$ In this reaction also, the electrophile (i.e. $\mathrm{Br}^{+}$) seemed to approach the 4,5 -double bond preferentially from one side, since a single bromoacetate (17) was obtained in $70 \%$ yield. The $\beta$-orientation of bromine function was inferred from the 10 -methyl n.m.r. signal, which appeared at low field ( $\delta 1.62$ ). Compound (17) was converted into the epoxide (18) by treatment with $10 \%$ sodium hydroxide in dimethylformamide. Similarly the dienone (19) was transformed, via the bromoacetate (21), into the epoxide (22).

Configurational assignments of the epoxides (16), (18), (20), and (22) were based on a comparative n.m.r. analysis (Table). The $\mathrm{C}-4$ proton signal in each case appeared as a doublet of doublets, suggesting a quasiequatorial orientation, since only this conformation is compatible with W -type coupling with the $\mathrm{C}-2$ proton. The C-10 methyl signal of the $\beta$-epoxide (16) appeared at $\delta 1.51$, to low field of that of the $\alpha$-isomer (18) ( $\delta 1.37$ ); this may be ascribable to deshielding by the 1 -oxo-function of (16). Analogously, compound (20) showed a significant lower field shift in comparison with its counterpart (22). Furthermore, the C-4 proton of the acetyl derivative of (18) resonated at $\delta 3.70$, owing to deshielding by the acetyl carbonyl group. The corresponding signal of the $\beta$-epoxide (16) acetate appeared at $\delta 3.30$.

Although participation ${ }^{13}$ by the neighbouring $6 \beta-$ hydroxy-group may assist in controlling the stereochemical course of the epoxidation of (15), the stereoselective

[^2]$4 \beta, 5 \beta$-attack seemed to be intrinsic in electrophilic reactions of the 2,4 -dien-1-one system. Thus, the $6 \alpha$ -hydroxy-analogue (19), as well as cholesta-2,4-dien-1-

(A)
one (26), showed a similar stereoselectivity in peroxyacid oxidation, bromoacetoxylation, and osmium tetraoxide oxidation (see later). Complete rationalization is difficult at present, but a molecular model indicates that the 2,4-diene-1-one system has a conformation similar to that of the 1,4 -dien-3-one system, ${ }^{14}$ in which rings $A$ and B may assume a quasi-cis and thence the 10 -methyl group a quasiequatorial conformation [see (A)]. In those situations, $\alpha$-side approach of reagents to the 4,5double bond may suffer from steric interaction with the $7 \alpha$ - and $9 \alpha$-hydrogen atoms.

Oxidation of the hydroxy-dienone (19) acetate with osmium tetraoxide provided the glycol (23) in $95 \%$ yield. The n.m.r. signal at $\delta 1.27$ due to the 10 -methyl group was shifted to $\delta 1.38$ by changing the solvent from $\mathrm{CDCl}_{3}$ to $\mathrm{C}_{6} \mathrm{D}_{6}$, and the $\mathrm{C}-4$ proton signal at $\delta 4.80$ showed coupling with the allylic C-2 proton ( $J 2 \mathrm{~Hz}$ ). These data strongly indicated a $4 \beta, 5 \beta$-structure for the glycol (23). In view of this information, a similar oxidation was carried out on the dienone (19) methanesulphonate, leading to the $4 \beta, 5 \beta$-glycol (24). This was directly treated with $15 \%$ sodium hydroxide in pyridine to afford the epoxide (25) in $85 \%$ yield from (19). The relevant n.m.r. signals (Table) showed perfect agreement with those of withaferin $\mathrm{A}(1)$, providing an additional proof of structure.

The remaining two analogues (39) and (40) were prepared from cholesta-2,4-dien-1-one (26). ${ }^{6}$ A stereoselective epoxidation of the dienone (26) with $m$-chloroperbenzoic acid gave the $4 \beta, 5 \beta$-epoxide (27) in $80 \%$ yield. The isomeric $4 \alpha, 5 \alpha$-epoxide (29) was also synthesized through bromoacetoxylation of (26), in a manner similar to that described above. Their stereochemistry was deduced from analogous considerations of n.m.r. spectra (Table), as for the structural assignments of compounds (16), (18), (20), and (22).

When the $\beta$-epoxide (27) was treated with $60 \%$ perchloric acid in tetrahydrofuran, the $4 \alpha, 5 \beta$-diol was obtained in $70 \%$ yield, and was characterized as its acetate (31). The presence of the $4 \alpha$-acetoxy-group was evidenced by conversion into the $4 \alpha$-acetoxy-dienone (33). Thus, dehydration of (31) with thionyl chloride in pyridine afforded three products. One of them ( $30 \%$ ) was identified as the enol acetate (32) from spectroscopic

[^3]evidence (Experimental section), and the other ( $50 \%$ ) was assigned structure (33), showing spin coupling between the 2 - and $4 \beta$-protons ( $J 2 \mathrm{~Hz}$ ) and also between the 3 and $4 \beta$-protons ( $J 2 \mathrm{~Hz}$ ). The third product (34) ( $15 \%$ )
attempted with phophorus pentaoxide in benzene; this gave the 1,4-dioxo-compound (35) ${ }^{15}$ ( $12 \%$ ) and the diacetoxy-compound (36) ( $30 \%$ ). The n.m.r. spectrum of (36) revealed two acetyl methyl signals at $\delta 1.95$ and
${ }^{1} \mathrm{H}$ N.m.r. ( $\delta$ values; $J$ and $W_{\frac{1}{2}}$ in Hz )

| Compound | 2-H | 3-H | 4-H | $J_{2.3}$ | $J_{2,4}$ | $J_{8.4}$ | 6-H | $W_{ \pm} 6-\mathrm{H}$ | $10-\mathrm{Me}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| (16) | 6.02 | 7.02 | 3.27 | 10 | 2 | 4 | 3.58 | 5 | 1.51 |
| (18) | 5.97 | 6.93 | 3.33 | 10 | 2 | 4 | 3.33 | 5 | 1.37 |
| (20) | 6.03 | 7.05 | 3.72 | 10 | 2 | 4 | 4.20 | 12 | 1.30 |
| (22) | 6.00 | 6.92 | 3.75 | 10 | 2 | 4 | 3.95 | 12 | 1.18 |
| (27) | 6.00 | 7.08 | 3.25 | 10 | 2 | 4 |  |  | 1.33 |
| (29) | 5.95 | 6.90 | 3.20 | 10 | 2 | 4 |  |  | 1.26 |
| (1) ${ }^{a}$ | 6.20 | 6.93 | 3.74 | 10 | 0 | 6 | 3.20 | 4 | 1.40 |
| (25) | 6.20 | 6.93 | 3.74 | 10 | 0 | 6 | 3.20 | 4 | 1.40 |
| (30) | 5.97 | 6.68 | 3.70 | 10 | 0 | 4 | 3.23 |  | 1.50 |
| (40) | 6.01 | 6.82 | 4.69 | 10 | 2 | 2 | 3.66 | 4 | 1.25 |
| (39) | 5.97 | 6.67 | 4.69 | 10 | 2 | 2 | 3.97 |  | 1.35 |
| (12) | 6.00 | 6.80 | $2.95{ }^{\text {b }}$ | 10 | 2 | 2 and 5 | 3.07 |  | 1.23 |
| (11) | 5.94 | 6.68 | $3.05{ }^{\text {b }}$ | 10 | 2 | 2 and 5 | 3.02 |  | 1.33 |

${ }^{a}$ We thank Dr. S. M. Kupchan for the spectrum of withaferin A (1). ${ }^{b}$ Only the $4 \beta$-proton signal was discerned; the $4 \alpha-\mathrm{H}$ signal was hidden in the methylene envelope.


Scheme 3 Reagents: i, $m$-CPBA; ii, NBA-AcOLi; iii, NaOH ; iv, $\mathrm{HClO}_{4}, \mathrm{Ac}_{2} \mathrm{O}, \mathrm{SOCl}_{2}, m$ - CPBA , and NaOH ; v, $\mathrm{HClO}_{4}$; vi, $\mathrm{Ac}_{2} \mathrm{O}$; vii, $\mathrm{POCl}_{3}$; viii, $\mathrm{P}_{2} \mathrm{O}_{5}$; ix, NaOH
might be produced in an $S_{\mathrm{N}} i$ fashion; its structure was deduced from its n.m.r. spectrum which is almost identical with that of the $5 \beta$-bromo-analogue (28). An unusual reaction was observed when dehydration of (31) was
2.18 and a signal corresponding to an equatorial proton (at C-4), $\alpha$ to the acetoxy-group ( $\delta 6.33, J_{3.4} 4 \mathrm{~Hz}$ ). ${ }^{15}$ E. Glotter, M. Weissenberg, and D. Lavie, Tetrahedron, 1970, 26, 3857.

Further corroboration came from identification with the product of acetylation of (31) with acetic anhydride-toluene-p-sulphonic acid.

The stereochemical course of the acid-catalysing ring opening of the epoxide (27) seemed to be controlled mainly by steric factors: hydroxide anion attacks the less hindered $\mathrm{C}-4$ in $S_{\mathrm{N}} 2$ fashion, rather than a tertiary C-5 carbocation. A similar situation was observed in bromoacetoxylation of the dienones (15), (19), and (26), where the intermediate bromonium ions appeared to be attacked by acetate ion regioselectively at C-4 in the anti sense.

Oxidation of the dienone acetate (33) with $m$-chloroperbenzoic acid afforded a mixture of two epoxides (37) and (38) in the ratio $3: 1$. Subsequent saponification yielded the corresponding alcohols (39) and (40). Their stereochemistry was deduced by comparing n.m.r. data (Table) with those of the 4-deoxy-analogues (11) and (12). The signals due to the 6 -proton in the $\beta$-epoxides (12) and (40) were doublets, whereas the corresponding signals of the $\alpha$-epoxides (11) and (39) were broad singlets. Furthermore the $\beta$-epoxides (12) and (40) showed 10 methyl signals at significantly higher field ( $\Delta \delta c a .0 .1$ p.p.m.) than the $\alpha$-epoxides (11) and (39).

## EXPERIMENTAL

M.p.s were determined with a hot-stage microscope. I.r. spectra were taken for KBr discs with a Hitachi ESI-G ${ }_{2}$ spectrometer and optical rotations for solutions in chloroform with a JASCO-DIP-S polarimeter. U.v. spectra were recorded by use of a Hitachi ESP-3T apparatus, with ethanol as solvent. ${ }^{1} \mathrm{H}$ N.m.r. spectra were run with a Varian T60 or JEOL JNM-4H-100 spectrometer $\left({ }^{2} \mathrm{H}\right)$ chloroform as solvent and $\mathrm{Me}_{4} \mathrm{Si}$ as internal reference). Mass spectra were determined with an LKB-9000S or Hitachi RMU-7L instrument at 70 eV .

Column chromatography was normally effected with Wako silica gel C-200. Preparative t.l.c. was carried out on $20 \times$ 20 cm plates, precoated with Merck silica gel $\mathrm{F}_{254}(2 \mathrm{~mm}$ thick). Organic extracts were dried over $\mathrm{Na}_{2} \mathrm{SO}_{4}$. 'The usual work-up' refers to dilution with water, extraction with an organic solvent, washing to neutrality, drying, filtration, and evaporation under vacuum. Ether refers to diethyl ether, THF to tetrahydrofuran, DMF to dimethylformamide, NBA to $N$-bromoacetamide, and $m$-CPBA to $m$-chloroperbenzoic acid.
$5 \alpha$-Cholest-2-ene-1 $\alpha, 6 \beta$-diol 6-Acetate (3).-The epoxyacetate (2) ( $24.7 \mathrm{~g}, 54 \mathrm{mmol}$ ) in propan-2-ol ( 520 ml ) was refluxed with hydrazine hydrate ( 130 ml ) and acetic acid $(6.5 \mathrm{ml})$. The usual work-up, with ethyl acetate for extraction, gave a crude product ( 32 g ). Chromatography on silica ( 300 g ) [benzene-hexane ( $4: 1$ )] afforded the allylic alcohol (3) ( $12 \mathrm{~g}, 48 \%$ ), m.p. $108-\mathrm{l} 10^{\circ}$ (from methanol) $[\alpha]_{\mathrm{D}}+74^{\circ}(c 1.72), \nu_{\text {max }} 3350,1735,1640$, and $1240 \mathrm{~cm}^{-1}$, $\delta 2.05(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 3.45(\mathrm{l} \mathrm{H}, \mathrm{s}, 1 \beta-\mathrm{H}), 5.05(\mathrm{lH}, \mathrm{m}, 6 \alpha-\mathrm{H})$, and $5.75(2 \mathrm{H}, \mathrm{s}, 2$ - and $3-\mathrm{H}$ ) (Found: C, $78.5 ; \mathrm{H}, 11.0$. $\mathrm{C}_{29} \mathrm{H}_{48} \mathrm{O}_{3}$ requires $\mathrm{C}, 78.3 ; \mathrm{H}, 10.9 \%$ ).
$6 \beta$-Acetoxy-5 $\alpha$-cholest-2-en-1-one (4).—The allylic alcohol (3) $(50 \mathrm{mg}, 0.11 \mathrm{mmol})$ in acetone ( 2 ml ) was treated with Jones reagent until the mixture turned orange-brown. The
usual work-up (ether for extraction) provided the crystalline enone (4) ( 50 mg ), m.p. $94-94.5^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}+74^{\circ}$ (c 1.55 ), $\lambda_{\text {max }} 227 \mathrm{~nm}, \nu_{\text {max }} 1735,1680$, and $1250 \mathrm{~cm}^{-1}, \delta$ $2.05(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 5.05(1 \mathrm{H}, \mathrm{m}, 6 \alpha-\mathrm{H}), 5.69(1 \mathrm{H}, \mathrm{m}, 2-\mathrm{H})$, and $6.63(\mathrm{l} \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$ (Found: $\mathrm{C}, 78.4 ; \mathrm{H}, 10.45 . \mathrm{C}_{29} \mathrm{H}_{46} \mathrm{O}_{3}$ requires $\mathrm{C}, \mathbf{7 8 . 7} ; \mathrm{H}, \mathbf{1 0 . 4 5} \%$ ).

Cholest-5-ene-1 $\alpha, 3 \beta$-diol 3 -Acetate (8).- $1 \alpha$-Hydroxycholesterol (7) ( $316 \mathrm{mg}, 0.79 \mathrm{mmol}$ ) was stirred with acetic acid ( 1.1 ml ) and acetic anhydride ( $0.5 \mathrm{ml}, 5.3 \mathrm{mmol}$ ) at $80^{\circ} \mathrm{C}$ for 4 h . The usual work-up (ether for extraction) gave a crude product ( 343 mg ). Chromatography on silica ( 17 g ) [ben-zene-ether ( $19: 1$ )] gave the monoacetate (8) ( $206 \mathrm{mg}, 59 \%$ ), m.p. $170-172^{\circ}$ (from methanol), $\delta 1.02(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.0$ ( $3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}$ ), $3.85(1 \mathrm{H}, \mathrm{m}, \mathrm{l} \beta-\mathrm{H}), 5.00(\mathrm{l} \mathrm{H}, \mathrm{m}, 3 \alpha-\mathrm{H})$, and 5.58 ( $1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}$ ). Elution with benzene-ether ( $\mathrm{I}: \mathrm{I}$ ) afforded unchanged (7) ( 123 mg ).
$6 \beta$-Hydroxy-5 $\alpha$-cholest-2-en-1-one (5).-The enone (4) $(5 \mathrm{~g}, 11.3 \mathrm{mmol})$ in dioxan ( 100 ml ) was refluxed with aqueous $15 \%$ sodium hydroxide $(34 \mathrm{ml})$ for 12 h . The usual work-up (ether for extraction) afforded the hydroxy-enone (5) $(4.5 \mathrm{~g}), \mathrm{m} . \mathrm{p} .145-147.5^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}+69^{\circ}$ (c $0.91)$, $\nu_{\max } 3420$ and $1660 \mathrm{~cm}^{-1}, \delta 1.23(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me})$, $3.95(1 \mathrm{H}, \mathrm{m}, 6 \alpha-\mathrm{H}), 5.80(1 \mathrm{H}, \mathrm{dd}, J 10$ and $3 \mathrm{~Hz}, 2-\mathrm{H})$, and $6.80(1 \mathrm{H}, \mathrm{m}, 3-\mathrm{H})$ (Found: $M^{+}, 400.3295 . \mathrm{C}_{27} \mathrm{H}_{44} \mathrm{O}_{2}$ requires $M, 400.3341$ ).
$3 \beta$-Acetoxycholest-5-en-1-one (9).-The $3 \beta$-monoacetate (8) $(206 \mathrm{mg}, 0.465 \mathrm{mmol})$ in acetone ( 10 ml ) was treated with Jones reagent. The usual work-up (ether for extraction) gave the crystalline ketone (9) ( 196 mg ), m.p. $142-144^{\circ}$ (from methanol), $\delta 1.28(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.05(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac})$, $5.0(\mathrm{l} \mathrm{H}, \mathrm{m}, 3 \alpha-\mathrm{H})$, and $5.58(\mathrm{l} \mathrm{H}, \mathrm{m}, 6-\mathrm{H})$ (Found: $M^{+}$, 442.3502. $\mathrm{C}_{29} \mathrm{H}_{46} \mathrm{O}_{3}$ requires $M, 442.3449$ ).

Cholesta-2,5-dien-1-one (10).-(a) The $6 \beta$-hydroxy-enone (5) $(1 \mathrm{~g}, 2.5 \mathrm{mmol})$ in pyridine $(20 \mathrm{ml})$ at $0^{\circ} \mathrm{C}$ was treated with phosphoryl chloride ( 1.1 ml ) for 3 h . The usual workup (ether for extraction) gave the dienone (10) ( 910 mg ), m.p. 98.5- $100^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}-18^{\circ}$ (c 1.49), $\delta 1.22$ $(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.95(1 \mathrm{H}, \mathrm{m}, 4 \alpha-\mathrm{H}), 3.20(\mathrm{lH}, \mathrm{m}, 4 \beta-\mathrm{H}), 5.60$ ( $1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}$ ), $5.85(1 \mathrm{H}, \mathrm{dq}, J 10.3$ and $2 \mathrm{~Hz}, 2-\mathrm{H}$ ), and 6.75 ( $1 \mathrm{H}, \mathrm{dq}, J 10.5$ and $3 \mathrm{~Hz}, 3-\mathrm{H}$ ) (Found: $M^{+}, 382.322$. $\mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}$ requires $M, 382.323$ ).
(b) The $3 \beta$-acetoxy enone (9) ( $60 \mathrm{mg}, 0.316 \mathrm{mmol}$ ) in dioxan ( 5 ml ) containing aqueous $15 \%$ sodium hydroxide $(0.5 \mathrm{ml})$ was stirred at room temperature for 6 h . The usual work-up (ether for extraction) gave the dienone (10) (55 mg ).
$5,6 \alpha-$ and $5,6 \beta$-Epoxy- $5 \alpha$ - and $-5 \beta$-cholest-2-en-1-ones (11) and (12).-The 2,5-dienone (10) ( $4.08 \mathrm{~g}, 10.6 \mathrm{mmol}$ ) in chloroform ( 187 ml ) was treated with $m$-CPBA for 3 h . The usual work-up provided a crude product ( 4.58 g ) containing two epoxides. Chromatography on silica ( 180 g ) [benzenehexane (4:1)] gave the $5,6 \beta$-epoxide ( 12 ) ( $1.27 \mathrm{~g}, 30 \%$ ), m.p. $113-115^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}+45^{\circ}(c 1.51)$ (Found: C, $81.45 ; \mathrm{H}, 10.8 . \mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{2}$ requires $\mathrm{C}, 81.35 ; \mathrm{H}, 10.6 \%$ ). Further elution with benzene-hexane ( $9: 1$ ) gave the $5,6 \alpha-$ epoxide (ll) $(2.40 \mathrm{~g}, 60.5 \%)$, m.p. $123-124^{\circ}$ (from methanol).

5,6 $\beta$-Dihydroxy-5 $\alpha$-cholest-2-en-1-one (13).-The $5,6 \alpha$ epoxide (11) ( $200 \mathrm{mg}, 0.50 \mathrm{mmol}$ ) in THF ( 27 ml ) was stirred with aqueous $3 \%$ perchloric acid ( 10.7 ml ) at room temperature for 5 h . The usual work-up (ether for extraction) gave the crystalline diol (13) ( 178 mg ), m.p. $141.5-143.5^{\circ}$ (from hexane-acetone), $[\alpha]_{\mathrm{d}}+37^{\circ}(c 1.53), \delta 1.32(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me})$, $3.30(1 \mathrm{H}, \mathrm{dt}, J 20$ and $3 \mathrm{~Hz}, 4 \beta-\mathrm{H}), 3.64(1 \mathrm{H}, \mathrm{m}, 6 \alpha-\mathrm{H}), 5.88$ ( $1 \mathrm{H}, \mathrm{dd}, J 10$ and $3 \mathrm{~Hz}, 2-\mathrm{H}$ ), and $6.60(1 \mathrm{H}, \mathrm{dq}, J 10$ and $3 \mathrm{~Hz}, 3-\mathrm{H}$ ) (Found: $M^{+}, 416.3248 . \mathrm{C}_{27} \mathrm{H}_{44} \mathrm{O}_{3}$ requires $M$,
416.3288). By the same method, the same diol ( 154 mg ) was obtained from the $5,6 \beta$-epoxide (12) ( 154 mg ).
$6 \beta-$ A cetoxycholesta-2,4-dien-1-one (14).-(a) A solution (3 ml) in pyridine of $6 \beta$-acetoxy- 5 -hydroxy- $5 \alpha$-cholest- 2 -en-1-one ( $44 \mathrm{mg}, 0.1 \mathrm{~mole}$ ), obtained by acetylation (acetic anhydridepyridine, room temperature, overnight) of the $5 \alpha, 6 \beta$-diol (13), was treated with thionyl chloride ( 0.09 ml ) at $0{ }^{\circ} \mathrm{C}$ for 20 min . The usual work-up (ether for extraction) gave the crystalline dienone ( 14 ) ( 40 mg ), m.p. $103-104.5^{\circ}$ (from methanol) $\lambda_{\text {max. }} 310 \mathrm{~nm}, v_{\text {max. }} 1735,1650,1620$, and 1240 $\mathrm{cm}^{-1}, \delta 2.0(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 5.52(\mathrm{IH}, \mathrm{m}, 6 \alpha-\mathrm{H}), 6.03(1 \mathrm{H}, \mathrm{d}, J$ $10 \mathrm{~Hz}, 2-\mathrm{H}), 6.3(\mathrm{l} \mathrm{H}, \mathrm{d}, J 8 \mathrm{~Hz}, 4-\mathrm{H})$, and $6.9(1 \mathrm{H}, \mathrm{dd}$, $J 10$ and $8 \mathrm{~Hz}, 3-\mathrm{H}$ ) (Found: $M^{+}-\mathrm{AcOH}, 380.3096$. $\mathrm{C}_{27} \mathrm{H}_{30} \mathrm{O}$ requires $M-\mathrm{AcOH}, 380.3077$ ).
(b) The $6 \beta$-acetoxy-enone ( 4 ) ( $1 \mathrm{~g}, 2.49 \mathrm{mmol}$ ) in carbon tetrachloride ( 15 ml ) was stirred with N -bromosuccinimide ( $600 \mathrm{mg}, 3.37 \mathrm{mmol}$ ) in sunlight for 3 h . After filtration, the solvent was evaporated off under vacuum. Chromatography on silica ( 50 g ) [hexane-benzene ( $1: 5$ )] gave the $4 \xi$-bromide ( $950 \mathrm{mg}, 73.5 \%$ ), m.p. $140-140.5^{\circ}$ (from methanol) $\nu_{\text {max }} 1740,1680$, and $1240 \mathrm{~cm}^{-1}$. The $4 \xi$-bromide $(950 \mathrm{mg}, 1.83 \mathrm{mmol})$ in dimethylacetamide ( 2 ml ) was added to a refluxing suspension of calcium carbonate ( 1 g ) in dimethylacetamide ( 10 ml ), and the mixture was refluxed for 30 min . The usual work-up (ether for extraction) gave a crude product ( 0.75 g ). Chromatography on silica ( 24 g ) [benzene-ether ( $9: 1$ )] afforded the dienone (14) ( 550 mg , 68\%).
$6 \beta$-Hydroxycholesta-2,4-dien-1-one (15).-The $5 \beta, 6 \beta$-epoxyenone (12) ( $1.26 \mathrm{~g}, 2.79 \mathrm{mmol}$ ) in dioxan ( 22 ml ) was refluxed with aqueous $15 \%$ sodium hydroxide $(2.52 \mathrm{ml})$ for 4 $h$. The usual work-up (ether for extraction) provided the crystalline dienone (15) ( 1.27 g ), m.p. $165-167^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}-146^{\circ}(c 1.55), \lambda_{\text {max. }} 312 \mathrm{~nm}, \nu_{\text {max }} 3475$, 1650 , and $1565 \mathrm{~cm}^{-1}, \delta 1.42(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 4.6(1 \mathrm{H}, \mathrm{m}$, $6 \alpha-\mathrm{H}), 5.9(1 \mathrm{H}, \mathrm{d}, J 9.6 \mathrm{~Hz}, 2-\mathrm{H}), 6.11(1 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 4-\mathrm{H})$, $6.9(\mathrm{l} \mathrm{H}, \mathrm{dd}, J 9.6$ and $6 \mathrm{~Hz}, 3-\mathrm{H})$ (Found: $M^{+}, 398.3132$. $\mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{2}$ requires $M$. 398.3182).
$4 \beta, 5$-Epoxy-6 $\beta$-hydroxy-5 $\beta$-cholest-2-en-1-one (16).-TThe $6 \beta$-hydroxy-dienone ( 15 ) ( $0.9 \mathrm{~g}, 2.26 \mathrm{mmol}$ ) in chloroform ( 40 ml ) was treated with $m-\mathrm{CPBA}(505 \mathrm{mg}, 2.49 \mathrm{mmol}$ ) at room temperature for 20 h . The usual work-up (chloroform for extraction) gave the crystalline epoxide (16) ( 963 mg ), m.p. 182.5-184.5 (from methanol), $[\alpha]_{\mathrm{D}}+11^{\circ}$ (c 1.55), $\lambda_{\text {max }} 242 \mathrm{~nm}, \nu_{\text {max }} 3500$ and $1650 \mathrm{~cm}^{-1}$ (Found: C, 78.45 ; $\mathrm{H}, 10.35 . \mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{3}$ requires $\mathrm{C}, 78.2 ; \mathrm{H}, 10.2 \%$ ).
$4 \alpha$-Aceto $x y-5$-bromo- $6 \beta$-hydroxy- $5 \beta$-cholest-2-en-1-one (17). A solution of the $6 \beta$-hydroxy-dienone ( 15 ) ( $220 \mathrm{mg}, 0.53$ mmol ) in acetic acid ( 8 m .) containing lithium acetate ( 800 $\mathrm{mg}, 12.1 \mathrm{mmol}$ ) and NBA ( $80 \mathrm{mg}, 0.58 \mathrm{mmol}$ ) was stirred at room temperature for 30 min . Water was added and the precipitate ( 261 mg ) was filtered off and crystallized from methanol, to afford the bromo-acetate (17), m.p. 182.5-$183^{\circ},[\alpha]_{\mathrm{D}}-128^{\circ}(c 1.55), \lambda_{\text {max }} 220 \mathrm{~nm}, \nu_{\text {max. }} 3400,1740$, 1665 , and $1230 \mathrm{~cm}^{-1}, \delta 1.62(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.17(3 \mathrm{H}, \mathrm{s}$, Ac), $4.0(1 \mathrm{H}, \mathrm{m}, 6 \alpha-\mathrm{H}), 6.13(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}, 2-\mathrm{H}), 6.18$ $(1 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 4 \beta-\mathrm{H})$, and $6.70(1 \mathrm{H}, \mathrm{dd}, J 10$ and 4 Hz , $3-\mathrm{H}$ ), $m / e 536$ and 538 ( $M^{+}$and $M^{+}+2$ ).
$4 \alpha, 5-E p o x y-6 \beta-h y d r o x y-5 \alpha$-cholest-2-en-1-one (18).—The bromo-acetate ( 17 ) ( $180 \mathrm{mg}, 0.355 \mathrm{mmol}$ ) in DMF ( 10.8 ml ) was treated with aqueous $10 \%$ sodium hydroxide at room temperature for 30 min . The usual work-up (chloroform for extraction) afforded a crude product ( 140 mg ). Chromatography on silica ( 6 g ) [benzene-acetone ( $100: 1$ )] gave the $\alpha$-epoxide (18) ( $70 \mathrm{mg}, 48 \%$ ), m.p. $184-187^{\circ}$ (from
hexane-acetone), $[\alpha]_{\mathrm{D}}+57^{\circ}$ (c 1.49) (Found: $M^{+}, 414.3200$. $\mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{8}$ requires $M$, 414.3132).
$6 \beta$-Hydroxycholesta-2,4-dien-1-one (19).-The $5 \alpha, 6 \alpha$ -epoxy-enone (li) ( $2.40 \mathrm{~g}, 5.28 \mathrm{mmol}$ ) in dioxan ( 42 ml ) was refluxed with aqueous $15 \%$ sodium hydroxide ( 4.8 ml ) for 4 h . The usual work-up (ether for extraction) gave the crystalline dienone (19) ( 2.27 g ), m.p. 156-157 ${ }^{\circ}$ (from methanol), $[\alpha]_{\mathrm{p}}-438^{\circ}(c \mathrm{l} .56), \lambda_{\text {max }} 320 \mathrm{~nm}, \delta 1.23(3 \mathrm{H}, \mathrm{s}$, $10-\mathrm{Me}), 4.53(\mathrm{l} \mathrm{H}, \mathrm{m}, 6 \beta-\mathrm{H}), 6.00(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}, 2-\mathrm{H}), 6.37$ ( $1 \mathrm{H}, \mathrm{d}, J 6 \mathrm{~Hz}, 4-\mathrm{H}$ ), and $7.03(1 \mathrm{H}, \mathrm{dd}, J 10$ and $6 \mathrm{~Hz}, 3-\mathrm{H}$ ) (Found: $\mathrm{C}, 8 \mathrm{I} .35 ; \mathrm{H}, 10.6 . \quad \mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{2}$ requires $\mathrm{C}, 81.35$; H, $10.6 \%$ ).
$4 \beta, 5-E p o x y-6 \alpha-h y d r o x y-5 \beta$-cholest-2-en-1-one (20).-The $6 \alpha$-hydroxy-dienone (19) ( $500 \mathrm{mg}, 1.26 \mathrm{mmol}$ ) in chloroform $(25 \mathrm{ml})$ was treated with $m$-CPBA ( $310 \mathrm{mg}, 1.52 \mathrm{mmol}$ ) at room temperature for 20 h . The usual work-up (chloroform for extraction) afforded the crystalline epoxide (20) ( 512 mg ), m.p. 150-151 ${ }^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}+3^{\circ}$ (c 1.59) (Found: $\mathrm{C}, 77.9 ; \mathrm{H}, 10.3 . \quad \mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{3}$ requires $\mathrm{C}, 78.2 ; \mathrm{H}, 10.2 \%$ ).
$4 \alpha$-A cetoxy-5-bromo-6 $\alpha$-hydroxy-5 $\beta$-cholest-2-en-1-one (21). -A solution of the $6 \alpha$-hydroxy-dienone (19) ( $660 \mathrm{mg}, 1.56$ mmol ) in acetic acid ( 24 ml ) containing lithium acetate $(2.4 \mathrm{~g})$ and NBA ( $240 \mathrm{mg}, 1.74 \mathrm{mmol}$ ) was stirred at room temperature for 40 min . The usual work-up (di-isopropyl ether for extraction) gave a crude product ( 851 mg ). Chromatography on silica ( 34 g ) [benzene-ether ( $10: 1$ )] afforded the bromo-acetate ( 21 ) ( $573 \mathrm{mg}, 67.5 \%$ oil), $\nu_{\text {max }} 3600,1735$, 1665 , and $1220 \mathrm{~cm}^{-1}, \delta 1.41(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), \stackrel{2}{\max } 10(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac})$, $4.40(\mathrm{l} \mathrm{H}, \mathrm{m}, 6 \beta-\mathrm{H}), 6.05(\mathrm{lH}, \mathrm{d}, J 10 \mathrm{~Hz}, 2-\mathrm{Hz}, 2-\mathrm{H}), 6.20$ ( $1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 4-\mathrm{H}$ ), and 6.66 ( $1 \mathrm{H} \mathrm{dd}, J 10$ and 5 Hz , 3-H).
$4 \alpha, 5$-Epoxy-6 $\alpha$-hydroxy-5 $\alpha$-cholest-2-en-1-one (22).-The bromo-acetate (21) ( $145 \mathrm{mg}, 0.313 \mathrm{mmol}$ ) in pyridine ( 5.8 ml ) was treated with N -sodium hydroxide ( 1.5 ml ) at room temperature for 30 min . The usual work-up (di-isopropyl ether for extraction) gave a crude product ( 113 mg ). Chromatography on silica ( 4.5 g ) [benzene-ether ( $50: 1$ )] afforded the $\alpha$-epoxide ( 22 ) ( $72 \mathrm{mg}, 60 \%$ ), m.p. 181-183 ${ }^{\circ}$ (from hexane-acetone), $[\alpha]_{\mathrm{D}}+66^{\circ}(c 1.51)$ (Found: $M^{+}, 414.3067$. $\mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{3}$ requires $M, 414.3132$ ).
$6 \alpha$-Acetoxy-4 $\beta, 5$-dihydroxy-5 $\beta$-cholest- 2 -en-1-one (23).-The $6 \alpha$-acetoxy-dienone ( $140 \mathrm{mg}, 0.315 \mathrm{mmol}$ ) [prepared from (19) by treatment with acetic anhydride-pyridine] in ether $(2 \mathrm{ml})$ was treated with pyridine ( 0.1 ml ) and osmium tetraoxide ( 0.34 mmol ) at room temperature for 30 h . The ether was evaporated off and the residue treated with pyridine ( 2 ml ) and saturated aqueous sodium hydrogen sulphite ( 2 ml ) at room temperature for 30 min . The usual work-up (ether for extraction) gave the 4,5-diol (23) (140 mg ), m.p. $17 \mathrm{I}-172^{\circ}$ (from methanol), $\delta 1.27(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me})$, $2.10(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 5.10(\mathrm{l} \mathrm{H}, \mathrm{m}, 4 \alpha-\mathrm{H}), 5.20(1 \mathrm{H}, \mathrm{m}, 6 \beta-\mathrm{H})$, $6.00(1 \mathrm{H}, \mathrm{dd}, J 10$ and $2 \mathrm{~Hz}, 2-\mathrm{H})$, and $6.57(1 \mathrm{H}, \mathrm{dd}, J 10$ and $2 \mathrm{~Hz}, 3-\mathrm{H}), \delta\left(\mathrm{C}_{6} \mathrm{H}_{6}\right) 1.37(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 1.67(3 \mathrm{H}, \mathrm{s}$, $\mathrm{Ac}), 4.80(\mathrm{lH}, \mathrm{m}, 4 \alpha-\mathrm{H}), 5.20(1 \mathrm{H}, \mathrm{m}, 6 \beta-\mathrm{H}), 5.93(\mathrm{l} \mathrm{H}$, dd, $J 10$ and $2 \mathrm{~Hz}, 2-\mathrm{H})$, and $6.32(1 \mathrm{H}, \mathrm{dd}, J 10$ and $2 \mathrm{~Hz}, 3-\mathrm{H}$ ).

5,6 $\beta$-Epoxy-4 3 -hydroxy-5 3 -cholest-2-en-1-one (25).-The $6 \alpha$-hydroxy-dienone (19) ( $113 \mathrm{mg}, 0.28 \mathrm{mmol}$ ) in pyridine ( 3.8 ml ) was treated with methanesulphonyl chloride ( 0.04 ml ) at $0^{\circ} \mathrm{C}$ overnight. The usual work-up (ether for extraction), performed below room temperature, gave the $6 \alpha$-mesylate ( 137 mg ). This product in ether ( 2 ml ) was treated with osmium tetraoxide ( $76 \mathrm{mg}, 0.30 \mathrm{mmol}$ ) and pyridine ( 0.07 ml ) at $0^{\circ} \mathrm{C}$ overnight. Ether was removed at room temperature. Pyridine ( 2 ml ) and saturated aqueous
sodium hydrogen sulphite were added to the residue, and the mixture was stirred at $0^{\circ} \mathrm{C}$ for 30 min . The usual work-up (ether for extraction) afforded the diol ( 128 mg ). This diol in pyridine ( 1.2 ml ) was treated with aqueous $15 \%$ sodium hydroxide $(0.075 \mathrm{ml})$ at $80^{\circ} \mathrm{C}$ for 15 min . The usual work-up (ether for extraction) provided the epoxide (25) ( 92 mg ), m.p. $220.5-223.5^{\circ}$ (from hexane-acetone), $[\alpha]_{\mathrm{D}}+39^{\circ}$ (c 1.60) (Found: $M^{+}, 414.3128 . \quad \mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{3}$ requires $M, 414.3132$ ).
$4 \beta, 5$-Epoxy-5 $\beta$-cholest-2-en-1-one (27).-The 2,4-dienone (26) $(75 \mathrm{mg}, 0.188 \mathrm{mmol})$ in chloroform ( 3.7 ml ) was treated with $m$-CPBA ( $41.2 \mathrm{mg}, 0.203 \mathrm{mmol}$ ) at room temperature for 20 h . The usual work-up gave the crystalline epoxide (27) ( 75 mg ), m.p. $155.5-158^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}+10^{\circ}$ (c 1.50 ), $\nu_{\text {max. }} 1660 \mathrm{~cm}^{-1}$ (Found: C, $81.5 ; \mathrm{H}, 10.7 . \mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{2}$ requires $\mathrm{C}, 81.35 ; \mathrm{H}, 10.6 \%$ ).
$4 \alpha$-Acetoxy-5-bromo-5 $\beta$-cholest-2-en-1-one (28).-The 2,4dienone (26) ( $550 \mathrm{mg}, 1.38 \mathrm{mmol}$ ) in acetic acid ( 20 ml ) was treated with lithium acetate ( $2 \mathrm{~g}, 30.3 \mathrm{mmol}$ ) and NBA ( 200 $\mathrm{mg}, 1.45 \mathrm{mmol}$ ) at room temperature for 30 min . The usual work-up (di-isopropyl ether for extraction) gave a crude product ( 750 mg ). Chromatography on silica ( 30 g ) [hexanebenzene ( $1: 1$ )] afforded the bromo-acetate (28) ( 500 mg , $69.5 \%$ ), m.p. $98-107^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}-132^{\circ}$ (c $1.55), \delta 1.44(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.16(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 6.08(1 \mathrm{H}, \mathrm{d}$, $J 10 \mathrm{~Hz}, 2-\mathrm{H}), 6.18(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 4 \beta-\mathrm{H})$, and $6.63(1 \mathrm{H}, \mathrm{dd}$, $J 10$ and $5 \mathrm{~Hz}, 3-\mathrm{H})$.
$4 \alpha, 5$-Epoxy- $5 \alpha$-cholest-2-en-1-one (29). -The bromo-acetate (28) ( $370 \mathrm{mg}, 0.71 \mathrm{mmol}$ ) in DMF ( 22 ml ) was treated with aqueous $10 \%$ sodium hydroxide ( 1.1 ml ) at room temperature for 1 h . The usual work-up (di-isopropyl ether for extraction) afforded a crude product ( 271 mg ). Chromatography on silica ( 11 g ) [benzene-hexane (3:2)] provided the $\alpha$ epoxide (29) ( $155 \mathrm{mg}, 55 \%$ ), m.p. $91-92^{\circ}$ (from methanol), $[\alpha]_{\mathrm{D}}+72^{\circ}$ (c 1.55) (Found: C, 81.5; H, 10.85. $\mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{2}$ requires $\mathrm{C}, 81.35 ; \mathrm{H}, \mathbf{1 0 . 6} \%$ ).
$5 \alpha, 6$-Epoxy-4 3 -hydroxy- $5 \alpha$-cholest-2-en-1-one (30).-The $4 \alpha, 5$-epoxy-enone (29) ( $60 \mathrm{mg}, 0.15 \mathrm{mmol}$ ) in THF ( 4.5 ml ) was treated with aqueous $60 \%$ perchloric acid ( 0.11 ml ) at room temperature for 3 h . The usual work-up (chloroform) afforded the diol ( 56 mg ), which was acetylated with acetic anhydride-pyridine. The resulting $4 \beta$-acetate ( 57 mg ) was treated with thionyl chloride ( 0.05 ml ) in pyridine ( 2.5 ml ) at $0{ }^{\circ} \mathrm{C}$ for 1 h . The usual work-up (ether) provided the 2,5 -dienone ( 48 mg ), $\delta 0.7(3 \mathrm{H}, \mathrm{s}, 13-\mathrm{Me}), 1.37(3 \mathrm{H}, \mathrm{s}, 10-$ Me ), $2.07(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 5.77(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 4 \alpha-\mathrm{H}), 6.03(1 \mathrm{H}$, d, $J 10 \mathrm{~Hz}, 2-\mathrm{H}), 6.12(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H})$, and $6.75(1 \mathrm{H}, \mathrm{dd}, J$ 10 and $5 \mathrm{~Hz}, 3-\mathrm{H})$.

The $4 \beta$-acetoxy-dienone ( 48 mg ) in chloroform ( 2.2 ml ) was treated with $m$-CPBA ( 30 mg ) at room temperature for 24 h . Preparative t.l.c. [benzene-ether ( $9: 1$ )] of the crude product ( 57 mg ) afforded $4 \beta$-acetoxy- $5,6 \alpha$-epoxy- $5 \alpha$ -cholest-2-en-l-one ( 24 mg ), $\delta 0.63(3 \mathrm{H}, \mathrm{s}, 13-\mathrm{Me}), \mathrm{l} .45(3 \mathrm{H}$, $\mathrm{s}, 10-\mathrm{Me}), 2.10(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 3.27(\mathrm{l} \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 6 \beta-\mathrm{H}), 4.80$ ( $1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}, 4 \alpha-\mathrm{H}$ ), $6.00(1 \mathrm{H}, \mathrm{d} J 10 \mathrm{~Hz}, 2-\mathrm{H})$, and $6.60(1 \mathrm{H}, \mathrm{dd}, J 10$ and $5 \mathrm{~Hz}, 3-\mathrm{H})$, and starting material ( 20 mg ). (In contrast to the results of Weissenberg et al., ${ }^{5}$ the isomeric $5 \beta, 6$-epoxide was not detected.) This epoxide ( 24 mg ) was treated with aqueous $10 \%$ sodium hydroxide $(0.06 \mathrm{ml})$ in dimethylformamide $(1.2 \mathrm{ml})$ at room temperature for 1 h . The usual work-up (ether) afforded the crystalline alcohol (30) ( 19 mg ) (for n.m.r. data see Table).
$4 \alpha$-Acetoxy-5-hydroxy-5 $\beta$-cholest-2-en-1-one (31).-The $4 \beta, 5 \beta$-epoxy-enone ( 27 ) ( $800 \mathrm{mg}, 2.01 \mathrm{mmol}$ ) in THF ( 60 ml ) was stirred with aqueous $60 \%$ perchloric acid ( 1.47 ml ) at
room temperature for 16 h . The usual work-up (ether for extraction) afforded the diol ( 820 mg ). Acetylation (acetic anhydride-pyridine, room temperature, overnight) yielded the crystalline monoacetate (31) ( 842 mg ), m.p. $169.5-173^{\circ}$ (from methanol), $v_{\max } 3450,1720,1670$, and $1260 \mathrm{~cm}^{-1}, \delta$ $1.22(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.17(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 5.53(\mathrm{l} \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}$, $4 \beta-\mathrm{H}), 6.07(\mathrm{IH}, \mathrm{d}, J 10 \mathrm{~Hz}, 2-\mathrm{H})$, and $6.60(\mathrm{lH}, \mathrm{dd}, J 10$ and $4 \mathrm{~Hz}, 3-\mathrm{H})$.

Dehydration of the Monoacetate (31) with Thionyl Chloride. -The monoacetate ( 31 ) ( $100 \mathrm{mg}, 0.218 \mathrm{mmol}$ ) in pyridine ( 5 ml ) was treated with thionyl chloride ( 0.06 ml ) at $0{ }^{\circ} \mathrm{C}$ for 1 h . The usual work-up (ether for extraction) provided a crude product,showing three spots on t.l.c. Each compound was isolated by preparative t.l.c: $4 \alpha$-aceto $x y-5$-chloro- $5 \beta$ -cholest-2-en-1-one (34) (upper band; $16 \mathrm{mg}, 15.4 \%$; oil), $\delta$ $1.39(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.18(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 5.97(1 \mathrm{H}, \mathrm{d}, J 5 \mathrm{~Hz}$, $4 \beta-\mathrm{H}), 6.10(\mathrm{I} \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}, 2-\mathrm{H})$, and $6.64(\mathrm{I} \mathrm{H}, \mathrm{dd}, J 10$ and $5 \mathrm{~Hz}, 3-\mathrm{H}$ ) (Found: $M^{+}, 476.3014 . \mathrm{C}_{29} \mathrm{H}_{45} \mathrm{ClO}_{3}$ requires $M$, 476.3057); $4 \alpha$-acetoxycholesta-2,5-dien-1-one (33) (middle band; $32 \mathrm{mg}, 33.3 \%$ ), m.p. $106-107^{\circ}$ (from methanol), $\nu_{\text {max }} 1740,1680$, and $1660 \mathrm{~cm}^{-1}, \delta 1.30(3 \mathrm{H}, \mathrm{s}$, $10-\mathrm{Me}), 2.20(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 5.67(1 \mathrm{H}, \mathrm{m}, 6-\mathrm{H}), 5.90(\mathrm{l} \mathrm{H}, \mathrm{dd}$, $J 10$ and $2 \mathrm{~Hz}, 2-\mathrm{H}), 6.20(1 \mathrm{H}, \mathrm{m}, 4 \beta-\mathrm{H})$, and $6.58(1 \mathrm{H}, \mathrm{dd}$, $J 10$ and $2 \mathrm{~Hz}, 3-\mathrm{H}$ ) (Found: C, 78.6; H, 9.95. $\mathrm{C}_{29} \mathrm{H}_{44} \mathrm{O}_{3}$ requires $\mathrm{C}, 79.05 ; \mathrm{H}, 10.05 \%$ ); $4 \alpha$-acetoxycholesta-2,4-dien-1-one (32) (lower band; $19 \mathrm{mg}, 19.8 \%$; oil), $\lambda_{\max } 322$ $\mathrm{nm}, \delta 1.30(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.22(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 5.98(\mathrm{lH}, \mathrm{d}, J$ $10 \mathrm{~Hz}, 2-\mathrm{H}$ ), and $6.77(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}, 3-\mathrm{H})$ (Found: $M^{+}$, 440.3260. $\quad \mathrm{C}_{29} \mathrm{H}_{44} \mathrm{O}_{3}$ requires $M, 440.3290$ ).

Dehydration of the Monoacetate (31) with Phosphorus Pentaoxide.-The monoacetate ( 31 ) ( $133 \mathrm{mg}, 0.3 \mathrm{mmol}$ ) in dry benzene ( 6 ml ) was treated with phosphorus pentaoxide ( $210 \mathrm{mg}, 14.4 \mathrm{mmol}$ ) at room temperature for 3 h . The usual work-up (ether for extraction) gave a crude product ( 98 mg ), showing three spots on t.l.c. Chromatography on silica ( 7 g ) [benzene-ether (5:1)] afforded cholest-2-ene-1,4-dione (35) ( $14 \mathrm{mg}, 12.3 \%$ ), $\nu_{\text {max }} 1680 \mathrm{~cm}^{-1}, \lambda_{\text {max }} 225 \mathrm{~nm}, \delta 1.25(3 \mathrm{H}, \mathrm{s}$, $10-\mathrm{Me})$ and $6.62(2 \mathrm{H}, \mathrm{s}, 2$ - and $3-\mathrm{H})$. Further elution with benzene-ether ( $5: 1$ ) gave the diacetate ( 36 ) ( $47 \mathrm{mg}, 29.1 \%$ ), $\delta 1.25(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 1.95(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 2.18(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 6.07$ $(1 \mathrm{H}, \mathrm{d}, J 10 \mathrm{~Hz}, 2-\mathrm{H}), 6.33(1 \mathrm{H}, \mathrm{d}, J 4 \mathrm{~Hz}, 4 \beta-\mathrm{H})$, and 6.67 ( $1 \mathrm{H}, \mathrm{dd}, J 10$ and $4 \mathrm{~Hz}, 3-\mathrm{H}$ ) (Found: $M^{+}, 500.3440$. $\mathrm{C}_{31} \mathrm{H}_{48} \mathrm{O}_{5}$ requires $M, 500.3501$ ). Continued elution with benzene-ether ( $5: 1$ ) afforded starting material (31) (27 mg ).
$4 \alpha$-Acetoxy-5,6 $\alpha$-epoxy- $5 \alpha$ - and $5 \beta$-cholest-2-en-1-ones (37) and (38).-The 2,5-dienone acetate (33) ( $107 \mathrm{mg}, 0.24 \mathrm{mmol}$ ) in chloroform ( 5 ml ) was treated with $m$-CPBA ( $63 \mathrm{mg}, 0.31$ mmol ) at room temperature for 6 h . The usual work-up afforded a crude product showing two spots on t.l.c. Each compound was isolated by preparative t.l.c. [benzene-ether ( $10: 1$ )]: the $\beta$-epoxide ( 38 ) $\left(20 \mathrm{mg}, 18.3 \%\right.$ ), m.p. $77-81^{\circ}$ (from ether), $\delta 1.27(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.08(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 3.52(\mathrm{l}$ $\mathrm{H}, \mathrm{d}, J 2 \mathrm{~Hz}, 6 \alpha-\mathrm{H}), 5.75(1 \mathrm{H}, \mathrm{m}, 4 \beta-\mathrm{H}), 6.05(1 \mathrm{H}, \mathrm{dd}, J$ 10 and $2.5 \mathrm{~Hz}, 2-\mathrm{H})$, and $6.65(\mathrm{l} \mathrm{H}, \mathrm{dd}, \mathrm{J} 10$ and $2 \mathrm{~Hz}, 3-\mathrm{H})$; the $\alpha$-epoxide ( 37 ) ( $75 \mathrm{mg}, 68.5 \%$ ), m.p. $161-165^{\circ}$ (from nhexane), $\delta 1.40(3 \mathrm{H}, \mathrm{s}, 10-\mathrm{Me}), 2.11(3 \mathrm{H}, \mathrm{s}, \mathrm{Ac}), 3.13(1 \mathrm{H}$, d, $J 5 \mathrm{~Hz}, 6 \beta-\mathrm{H}), 6.01(2 \mathrm{H}, \mathrm{m}, 2$ - and $4 \beta-\mathrm{H})$, and $6.55(1 \mathrm{H}$, dd, J 10 and $2 \mathrm{~Hz}, 3-\mathrm{H}$ ).
$5,6 \alpha$-Epoxy-4 $\alpha$-hydroxy- $5 \alpha$-cholest-2-en-1-one (39).-The $5 \alpha, 6 \alpha$-epoxy-acetate ( 37 ) ( $30 \mathrm{mg}, 0.066 \mathrm{mmol}$ ) in THF ( 1.5 ml ) was treated with 2 N -potassium hydroxide ( 0.21 ml ) at room temperature overnight. The usual work-up (ether for extraction) gave a crude product ( 29 mg ) which was purified by preparative t.l.c. [benzene-acetone (8:2)] to
afford the epoxy-alcohol (39), m.p. 195-200 (from ether) (Found: $M^{+}, 414.3120 . \quad \mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{3}$ requires $M, 414.3132$ ). $5,6 \beta-E p o x y-4 \alpha-h y d r o x y-5 \beta$-cholest-2-en-1-one (40)-Similar treatment of the $5,6 \beta$-epoxide ( 38 ) ( $10 \mathrm{mg}, 0.022 \mathrm{mmol}$ ) gave a crude product ( 10 mg ) which was purified by preparative t.l.c. [benzene-acetone ( $8: 2$ )] to afford the alcohol (40), m.p.

200-209 (from ether) (Found: $M^{+}, 414.3080 . \mathrm{C}_{27} \mathrm{H}_{42} \mathrm{O}_{3}$ requires $M, 414.3132)$.

This work was supported by a research grant from the Ministry of Health and Welfare.
[5/875 Received, 9th May, 1975]


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